O-Carbamoyl Oximes as Potential Analgesics

Gerald N. Evenson* and Robert Bruce Moffett

Research Laboratories, The Upjohn Company, Kalamazoo, Michigan 49001 Received August 31, 1979

Preparation of oxadiazinones 2, from α -anilinoacetophenone oximes and O-carbamoyl-2,3-dihydro-4-quinolinone oximes 3, from quinolinones are described.

J. Heterocyclic Chem., 17, 351 (1980).

In our continuing search for novel analysis agents, two types of compounds, 2 and 3, were prepared having the O-carbamoyl oxime structure characteristic of anidoxime (E-142) 1 (1).

Oxadiazinones, 2 (R = H) were made by a modification of the method of Gnichtel and Thiele (2) from α -anilinoacetophenone oximes (Scheme I).

Attempts to place an aminomethyl function (R = CH₂NR'₂) at the 4-position failed either by the Mannich reaction on 2 or by introducing it at an earlier stage.

A series of O-carbamoyl-2,3-dihydro-4-quinolinone oximes, 3, were prepared as shown in Scheme II.

The intermediate quinolinones leading to 3 were made by modifications of the method of Atwal, et al (3) from β -anilinopropionic acids and reductive alkylations of the unsubstituted quinolinones using acetaldehyde and sodium cyanoborohydride. The data for the O-carbamoyl-2,3-dihydro-4-quinolinones are listed in Table I.

All the novel compounds were tested for analgesia in mice but were inactive. However, screening in a battery of tests for central nervous system activity showed weak to moderate body temperature lowering in mice. This may be indicative of CNS depressant activity.

EXPERIMENTAL

Melting points were determined on a Thomas-Hoover apparatus. Calibration against standard compounds showed no correction was necessary.

Dihydro-5-(4-methoxyphenyl)-3-phenyl-6H-1,2,5-oxadiazin-6-one (2a).

A solution of 2-[(4-methoxyphenyl)amino]acetophenone oxime (12.8 g., 50.0 mmoles), prepared by the method of Julian, et al (4), in tetrahydrofuran (75 ml.) was treated with 1,1'-carbonyldiimidazole (9.7 g., 60.0 mmoles) and stirred at ambient temperature under nitrogen for 2.5 hours. The mixture was concentrated in vacuo and the residue dissolved in methylene chloride, washed with water, dried (sodium sulfate), and concentrated in vacuo. The oil was crystallized from ethyl acetate-hexane to give 3.95 g., m.p. 99-101.5°. The filtrate was chromatographed on silica gel eluting with chloroform to give 2.32 g., m.p. 98.5-100° of a white solid (overall yield 45%); ir (Nujol): 1734 (C=O), 1610, 1585, 1515, 1500 (C=C/C=N), 1245, 1140, 925 cm⁻¹ (C-O/C-N/N-O); nmr, (deuteriochloroform): δ 3.8 (s, 3H), 4.7 (s, 2H), 6.9 (d, 2H), 7.3 (d, 2H), 7.7 (m, 5H). Anal. Calcd. for C_{1.6}H₁₄N₂O₃: C, 68.07; H, 5.00; N, 9.92. Found: C, 67.89; H, 4.93; N, 9.67.

Dihydro-5(4-methoxyphenyl)-3(4-nitrophenyl)-6H-1,2,5-oxadiazin-6-one (2b).

A solution of p-nitrophenacyl-p-anisidine oxime (0.3 g., 1.0 mole), prepared by the manner of Julian, et al (4), in tetrahydrofuran (10 ml.) was treated with 1,1'-carbonyldiimidazole (0.24 g., 1.5 mmoles) and stirred at ambient temperature under nitrogen for 2.5 hours. The solution was concentrated in vacuo and the residue dissolved in methylene chloride, washed with water, dried (sodium sulfate), and concentrated in vacuo. The product was recrystallized from methylene chloridemethanol-ethyl acetate to give a white solid, m.p. 154-156°, 0.2 g. (61%); ir (Nujol): 1730 (C=O), 1605, 1530, 1515 (C=C/C=N/NO₂), 1345 (NO₂), 1250, 1145, 1035, 935 cm⁻¹ (C-O/C-N/N-O); nmr (deuteriochloroform): δ 3.83 (s, 3H), 4.8 (s, 2H), 6.95 (d, 2H), 7.35 (d, 2H), 7.9 (d, 2H), 8.3 (d, 2H).

Anal. Calcd. for C₁₆H₁₃N₃O₅: C, 58.72; H, 4.00; N, 12.84. Found: C,

58.96; H, 4.12; N, 12.78. © *HeteroCorporation*

26 (X = NO2, R = H)

Table I

O-Carbamoyl-2,3-dihydro-4-quinolinone Oximes

								Analysis						
Compound				%			Calcd.				Found			
No.	X	R	Y	M.p. °C	Yield	Formula	С	H	N	S/Cl	С	H	N	S/Cl
3a	Н	CH ₃	4-OCH ₃	133-137	41	$C_{25}H_{21}N_3O_6S$ (a)	60.35	5.47	8.45	6.44	60.45	5.40	8.40	6.53
3b	H	CH ₃	4-Cl	119-124	84	$C_{17}H_{16}ClN_3O_2$	61.91	4.89	12.74	10.75	61.86	4.92	12.78	10.79
3c	H	C_2H_5	H	159-162	88	$C_{18}H_{19}N_3O_2$	69.88	6.19	13.58		69.71	5.82	13.38	
3d	H	C_2H_5	4-OCH ₃	94-96.5	65	$C_{19}H_{21}N_3O_3$	67.24	6.24	12.38		67.21	6.34	12.50	
3 e	H	C ₂ H ₅	4-Cl	165-168	84	$C_{18}H_{18}CIN_3O_2$	62.88	5.28	12.22	10.31	67.76	5.45	12.21	10.26
3f	H	C_2H_5	2,5-DiCl	161-164	93	$C_{18}H_{17}Cl_2N_3O_2$	57.15	4.53	11.11	18.75	57.29	4.62	11.21	18.80
3g	H	C_2H_5	4-NO ₂	175-177 (b)	80	$C_{18}H_{18}N_{4}O_{4}$	61.01	5.12	15.81		60.62	5.44	15.94	
3h	6-OCH ₃	C ₂ H ₅	4-OCH ₃	139-144	37	$C_{20}H_{23}N_3O_4$	65.03	6.28	11.38		64.87	6.34	11.44	
3i	6-OCH ₃	C ₂ H ₅	4-Cl	131-135	66	$C_{19}H_{20}ClN_3O_3$	61.04	5.39	11.24	9.48	60.95	5.43	11.23	9.61
3j	6-OCH,	C,H,	2,5-DiCl	162-163.5	60	$C_{19}H_{19}Cl_2N_3O_3$	55.89	4.69	10.29	17.37	56.14	4.64	10.45	17.52
3k	6-OCH ₃	C ₂ H ₅	4-NO ₂	122-125	63	$C_{19}H_{20}N_4O_5$	59.37	5.24	14.58		59.33	5.28	14.58	

(a) Obtained as 4-methylbenzenesulfonate salt. (b) Had distinct softenings at 124°, 144° and 164°.

General Procedure for O-Carbamoyl-2,3-dihydro-4-quinolinone Oximes (3).

A solution of the appropriately substituted 2,3-dihydro-4-(4H)quinolinone oximes were dissolved in benzene and treated with an equivalent amount of the substituted phenylisocyanate. The mixtures were stirred at ambient temperature under nitrogen for 1 hour. The mixtures were concentrated in vacuo and the residues dissolved in methylene chloride, washed with water, dried (sodium sulfate), and concentrated in vacuo. The residues were crystallized from ethyl acetate-hexane to give the products listed in Table I.

Acknowledgement.

The authors wish to thank the Physical and Analytical Chemistry Unit

for analytical and spectral data, and Dr. Philip VonVoigtlander for the biological testing.

REFERENCES AND NOTES

- (1) J. R. Shroff, P. Cervoni, and M. H. Karten, 10th Great Lakes Regional ACS Meeting, June 18, 1976; D. J. Grainger, T. H. Gawley, and J. W. Dundee, *Br. J. Anaesth.*, 49, 257 (1977).
 - (2) H. Gnichtel and S. Thiele, Chem. Ber., 104, 1507 (1971).
- (3) M. S. Atwal, L. Bauer, S. N. Dixit, J. E. Gearien, and R. W. Morris, J. Med. Chem., 8, 566 (1965).
- (4) P. L. Julian, E. W. Meyer, A. Magnani, and W. Cole, J. Am. Chem. Soc., 67, 1203 (1945).